

# Recent Developments in Low-Temperature Solution-Processed Metal Oxide Electron Transport Layers for Perovskite Solar Cells

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## ABSTRACT

With power conversion efficiency (PCE) exceeding 25%, perovskite solar cells (PSC) have become known as a remarkable photovoltaic technology in the last few years. Further, the low preparation temperature and facile processing techniques specifically in the planar architecture have contributed immensely to the production of inexpensive solar devices. The charge transport layers, primarily the electron transport layer (ETL) play a significant role in efficient charge extraction and transport from the absorber layer to the respective electrode. In particular, numerous research have been carried out. On the oxides of metal (MOX) based ETL owing to their exceptional optoelectronic characteristics, low cost, and remarkable adaptability. This review deals with the application of newly developed MOX-based ETLs processed at low temperatures for planar PSC. The optical and electrical characteristics of widely used MOX ETL were briefly investigated and explored for impact on the photovoltaic (PV) stability and performance of halide PSC. In the end, the future perspective on the MOX ETL was also offered.

**Keywords:** Electron transport layer, metal oxides, optoelectronics, perovskite solar cell

## Introduction

The biggest challenge to mankind in the next 50 years will be related to the energy crisis and environmental disorder [1]. Our industrialized civilization has an ever-increasing demand for energy and the only viable solution is fossil fuels which has already led to the depletion of the earth's oil reserves [2]. The global world is suffering due to energy-related problems and the possible remedial measure is to shift the dependence on sustainable and green energy resources [3]. In our daily lives, we use a variety of energy resources, including oil, coal, solar, and wind energy. Due to non-renewable energy resources, different kinds of pollutants and hazardous gasses are released into the atmosphere [4]. Numerous health issues arise from the use of fossil fuels. In addition, it produces pollution of the water, land, and air. Furthermore, it raises the temperature of the earth's surface and the atmosphere resulting in severe climate changes [5]. Therefore, to resolve these issues, alternative energy resources must be explored. Solar, wind, and other forms of renewable energy resources are extensively investigated today to gain independence from fossil fuels in the future. One of the greatest sources of renewable energy is sun energy which is a limitless, environmentally benign, and cost-effective resource [6]. "In just one hour, the solar energy striking the planet is predominantly higher than the total energy utilized by humans in one year" [7]. The global energy demand can be overcome by the use of solar energy-based devices [8]. Numerous PV technologies have been discovered to date for effectively capturing solar energy and generating power. Perovskite solar cells, a developing superstar among them, are considered to be the most promising option due to their affordable fabrication cost, simplicity in manufacturing, and high power conversion efficiency (PCE) [9-11].

Through revolutionary advancement in the development of perovskite thin films with high crystalline nature, interfacial design, and device engineering, PSCs have attained a remarkable PCE of 26.5%, setting a new world record [12-14]. The rapid advancement in PCE of PSC is mainly attributed to exceptional optoelectrical characteristics of perovskite material like large carrier diffusion length, low

binding energy of exciton, greater light absorption capacity, and long carrier lifespans [15-17]. Miyasaka's team introduced MAPbBr<sub>3</sub> nanocrystalline thin-film as a sensitizer in a dye-sensitized solar cell (DSSC) in 2009 achieving a PCE of 3.8% [18]. However, these perovskite-sensitized solar cells exhibited bad photostability. Using electrolytes in liquid form critically degraded the perovskite films. Ever since Park et al. presented spiro-OMeTAD as a potential hole-carrying material and incredibly fabricated the first common solid-state PSC [19].

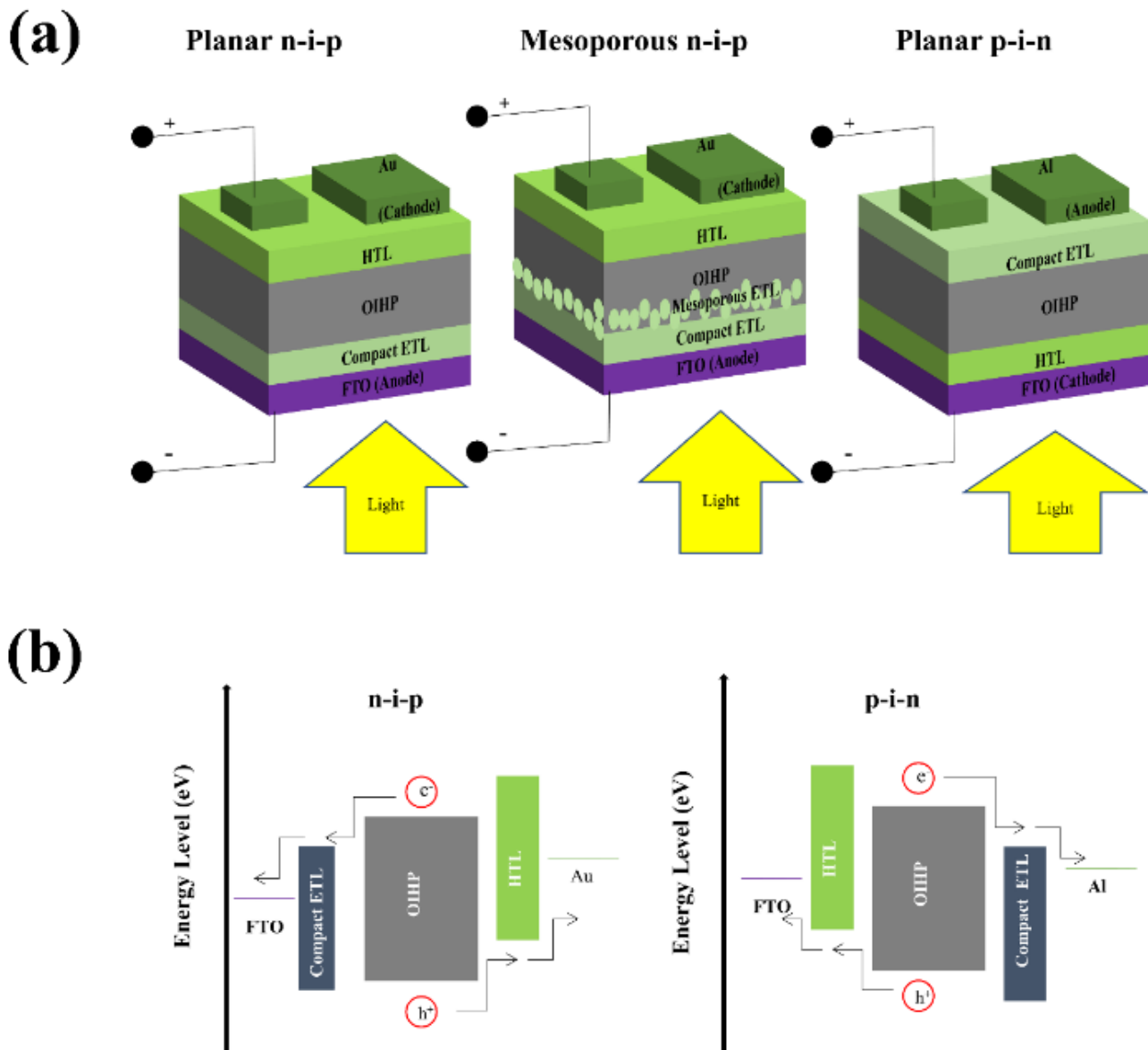
One of the most significant device architectures is the mesoporous solid-state perovskite solar cell. Unfortunately, the mesoporous TiO<sub>2</sub> often needs to be sintered at a high temperature. Of over 500 °C to enhance film characteristics and crystallinity [20]. High-temperature annealing and complex film formation processes limit its application in the fabrication of flexible devices [21]. The development of planar-type PSC can reduce the processing temperature and speed up the production process and is a distinctive technique for device fabrication with versatile compatibility [22-24]. Even in the absence of the mesoporous layer, the charge carriers (electrons and holes) can effectively travel to their respective electrodes due to the long carrier diffusion length and long carrier lifespan of the absorber material in planar architectures [25, 26]. The standard planar PSC can now be divided into two structural types: planar n-i-p (regular) and planar p-i-n (Inverted) illustrated in Figure 1a. In particular, ETLs are indispensable for the efficient collecting and transport of photogenerated electrons in planar PSC (Figure 1b), for blocking holes, and for reducing interfacial recombination [22-25]. In PSC, electron transport material (ETM) mostly utilized can be of organic or inorganic nature [27]. Organic ETLs excel in both flexibility and better solution processes [28]. Yet, they suffer from high costs and manifest deprived thermo-mechanical and environmental stability [29-31]. Inorganic n-type substances, particularly MO<sub>x</sub> like titanium dioxide, zinc oxide, tin oxide, and iron oxide are frequently employed ETLs owing to their superior optoelectronic properties, facile solution processed techniques, low annealing temperatures, tremendous adaptability, and high stability [22-24, 32, 33]. Currently, a

lot of research is being carried out to explore novel ETL materials and improve  $\text{MO}_x$  ETLs to expand their application in PSC [34-36]. Various reviews on the development and progress of  $\text{MO}_x$  ETLs in PSC have been published, and a few of them have systematically concentrated on the  $\text{MO}_x$  ETLs processed at low temperatures in planar PSC [21, 37-39]. In this review, the electrical and optical properties of the popular  $\text{MO}_x$  ETLs  $\text{TiO}_2$ ,  $\text{ZnO}$ ,  $\text{SnO}_2$ , and  $\alpha\text{-Fe}_2\text{O}_3$  are discussed, followed by the impact of elemental doping and thin film formation. The latest developments in low-temperature processing of these  $\text{MO}_x$  ETLs for planar PSC are then reviewed. Lastly, a viewpoint on the challenges and

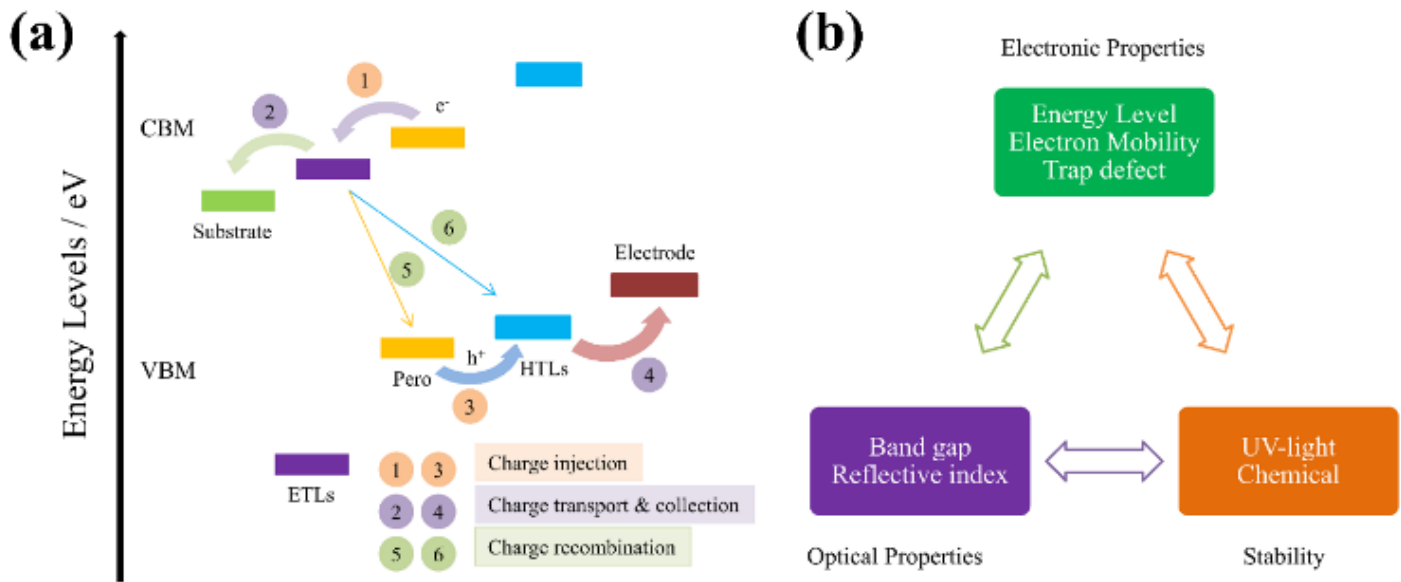
bright future of  $\text{MO}_x$  ETLs processed at low temperatures is presented.

### Metal Oxide as ETL Materials Fundamental Characteristics of $\text{MO}_x$ -based ETLs

The versatile optoelectronic properties of ETLs have an important impact on the photovoltaic solar cell performance. An important consideration is the energy level and the band gaps of the  $\text{MO}_x$ -based ETLs as shown in in Figs. 2a and 2b.



**Figure 1.** (a) Different configurations of a PSC including planar and mesoporous architectures; and (b) Energy level diagram of planar PSC.



**Figure 2.** (a) Diagrammatic illustration of the charge transfer mechanism in standard PSC based on  $\text{MO}_x$  ETLs; and (b) fundamental performance influencing parameters for optimal  $\text{MO}_x$  ETLs.

To facilitate smooth electron extraction by effectively blocking the unwanted hole carriers, a perfect alignment of the band between the  $\text{MO}_x$  ETLs and the perovskite thin layer is desired [40]. Furthermore, for effective charge transfer and collection by lowering interfacial recombination and interface contact resistance, a fast electron mobility of the ETL is required. Wide band gap and low refractive index  $\text{MO}_x$  typically have favorable transmittance, which could increase the amount of sunlight that reaches the perovskite absorption layer in a typical PSC. To ensure the durability of the contact between ETLs and perovskite,  $\text{MO}_x$  ETLs on conductive substrates should also be UV-insensitive [41]. To prevent gradual interaction with the neighboring conductive electrode or perovskite film under certain circumstances, a  $\text{MO}_x$  ETL must possess strong chemical stability [42]

Among n-type  $\text{MO}_x$ ,  $\text{TiO}_2$ ,  $\text{ZnO}$ ,  $\text{SnO}_2$  and  $\alpha\text{-Fe}_2\text{O}_3$  are preferred ETLs for superior performance planar PSC. Due to the broader bandgap and appropriate energy levels with the absorber layer,  $\text{TiO}_2$  is the most commonly used ETL in planar PSC and was previously utilized as a potential photoanode in DSSC [21]. The conduction band minimum (CBM) of  $\text{TiO}_2$  is specifically located at  $-4.1$  eV, with the band gaps for anatase, brookite, and rutile being  $3.2$  eV,  $3.1$  eV, and  $3.0$  eV, respectively [43, 44]. Moreover, it possesses a low index of refraction  $2.4\text{--}2.5$ . But,  $\text{TiO}_2$  suffers from low mobility of electrons from  $0.1\text{ cm}^2\text{V}^{-1}\text{s}^{-1}$  up to  $4\text{ cm}^2\text{V}^{-1}\text{s}^{-1}$  in bulk ranging [37]. The strong catalytic behavior of  $\text{TiO}_2$  may cause deterioration of the perovskite layer under ultraviolet (UV) light [45]. Another common ETM that has gained interest as a substitute for  $\text{TiO}_2$  is  $\text{ZnO}$ .  $\text{ZnO}$  is similar to  $\text{TiO}_2$  due to its photoactive nature and possesses a fair energy level structure and a high visible light transmittance. Further,  $\text{ZnO}$  has a bulk electron mobility that is significantly higher than  $\text{TiO}_2$  ( $> 200\text{ cm}^2\text{V}^{-1}\text{s}^{-1}$ ) [46-48].  $\text{ZnO}$  also exhibits good structural tunability at the nanoscale. It can be easily

crystallized to produce many kinds of nanocomposites at low temperatures, including nanoparticles, nanorods, and nanoflowers [37]. However, its application is restricted because of its poor chemical and thermal stability particularly at the  $\text{ZnO}$ /perovskite interface as a result of hydroxyl groups or left over acetate ligands on the  $\text{ZnO}$  surface [49].  $\text{SnO}_2$  has received substantial interest recently and is considered to be the most wonderful ETL due to its exceptionally high mobility of electrons and reasonably deep conduction band minimum ranging from  $-4.2$  eV to  $-4.5$  eV. Additionally,  $\text{SnO}_2$  possesses a high transmittance across the complete visible spectrum, a large optical band gap, and a tiny refractive index [38, 50]. In contrast to all other ETLs,  $\text{SnO}_2$  was notable for its low photo-catalytic activity, strong chemical stability, and resistance to UV light [50, 51]. The energy level alignment matching between  $\text{MO}_x$  ETL and the perovskite layer ensures efficient charge extraction without recombination loss. However, materials (e.g.,  $\text{SnO}_2$ ) with the conduction band level that can match the perovskite layer well but block hole transfer make it easy for electron injection into the perovskite layer and preventing a hole from injecting in from the anode side, thereby boosting device performance. Tin (Sn) scarcity of resources, however, is one of its main issues. According to a recent International Sn Association study, Sn availability is finite and advises researchers to search for an other metal. The study also demonstrates that Sn availability is already crucial in the US and China, it is about to become scarce in the EU as well [52]. Hematite ( $\alpha\text{-Fe}_2\text{O}_3$ ) is a thermodynamically stable n-type semiconducting iron oxide that has gained global attention due to its low UV photocatalytic activity that is useful for prolonging the UV stability of PSC [53]. However,  $\text{MO}_x$  still has some inherent issues like trap states, poor mobility, and low energy levels, despite their great potential for use as ETLs in PSC.

ETL	$E_g/eV$	CBM/eV	Refractive Index	Bulk Mobility ( $cm^2.V^{-1}.S^{-1}$ )	Ref.
TiO <sub>2</sub>	3.0-3.2	-4.1	2.4-2.5	0.1-4	[37, 43, 44]
ZnO	3.1-3.3	-4.17	2.1	205	[38, 46-48]
SnO <sub>2</sub>	3.6-4.0	-4.3	2.0	240	[37-39, 50]
WO <sub>x</sub>	2.6-3.1	-4.5	1.95	10-20	[54-56]
Nb <sub>2</sub> O <sub>5</sub>	3.2-3.9	-4.25	2.1-2.4	0.2-0.4	[57-59]
Cr <sub>2</sub> O <sub>3</sub>	3.5	-3.93	1.4-2.1	10 <sup>-5</sup> -1	[60-62]
CeO <sub>x</sub>	3.5	-4.0	1.6-2.5	0.01	[38, 63, 64]
Fe <sub>2</sub> O <sub>3</sub>	2.2	-4.5	2.7-3.5	0.01-0.1	[65-67]
Zn <sub>2</sub> SnO <sub>4</sub>	3.8	-4.1	2.0	10-30	[68, 69]
BaSnO <sub>3</sub>	3.1	-4.3	2.07	50	[70-72]
TiZn <sub>2</sub> O <sub>4</sub>	3.29	-4.13	-	-	[73]
Ti <sub>0.5</sub> Fe <sub>0.5</sub> O <sub>x</sub>	2.65	-4.08	-	-	[74]

**Table 1**

Comparison of electrical characteristics for binary and ternary MO<sub>x</sub> ETLs in planar Perovskite solar cell

Doping is widely used to adjust the physicochemical, optical, and electrical characteristics of MO<sub>x</sub> [75]. In general, this technique could be beneficial in shifting the CB maxima to improve energy level alignment as well as passivate oxygen vacancies and decrease surface trap states of MO<sub>x</sub> [49, 76]. Additionally, doping may also greatly improve electron transport and extraction from MO<sub>x</sub> and increase its mobility. Surface modification is another useful strategy for improving electrical characteristics and minimizing surface flaws in MO<sub>x</sub> [77, 78]. Furthermore, the development of new composites such as mixing MO<sub>x</sub> directly with carbon materials or with another binary MO<sub>x</sub> possessing prerequisite characteristics is beneficial in manipulating MO<sub>x</sub> properties through synergistic effects. For example, SnO<sub>2</sub> layers decorated with graphene quantum dots (GQDs) enhanced film uniformity and conductivity. By manipulating the quantity and composition of GQDs, higher fermi levels were achieved, which considerably enhanced the transfer of electrons and device efficiency of a conventional PSC [79]. The electrical properties of ternary and binary MO<sub>x</sub> ETLs that are often employed as ETL in planar PSC are shown in Table 1.

#### **Techniques for Deposition of MO<sub>x</sub> ETLs**

Low-temperature solution-processed MO<sub>x</sub> ETLs can be deposited on a perovskite absorber or conductive substrate in a PSC by a variety of methods. The properties of universal approaches are listed below.

#### **Solution Process Method**

Low-temperature ETLs are usually deposited using the solution-processed approach, which involves either directly producing MO<sub>x</sub> nano-crystals and nano-colloidal or by thermal decomposition of the precursors of metal salt as displayed in Figure 3a [80-83]. To deposit a metal salt precursor on the substrate via the thermal decomposition technique, the precursors of metal salt are initially mixed with

the suitable solvent and then deposited on glass conductive substrates by spin coating, spray coating, etc. Thermal annealing was then conducted to transform the allocated films into the desired MO<sub>x</sub> layers. The separation of the film formation and metal oxide crystallization enables direct coating of the resulting nano-sized solution using a range of conventional methods, including spin coating, slot die coating, and ink jet printing. In particular, the slot die coating and ink jet printing processes show great promise for the large-scale manufacturing of MO<sub>x</sub> ETLs. However, the solution processing method is still widely employed for various MO<sub>x</sub> ETLs, particularly processed at low temperatures for efficient planar PSC due to superior crystallinity and fewer defects in the as-deposited films [84]. It must be underlined that the annealing temperature and relative humidity are important factors in the production of good-quality MO<sub>x</sub> layers via the process of thermal decomposition [38, 50].

#### **Atomic Layer Deposition (ALD) Method**

Compact, homogenous, and conformal thin films have been developed by the semiconductor industry through the use of the ALD technique, a critical technology for precisely controlling film growth [85]. The film thickness may be easily controlled, and it even allows for the formation of films on textured substrates. The thickness of the MO<sub>x</sub> ETL plays an important role in the actual performance resulting in the device. An adequate thickness is sufficient to provide enough electron transport as well as to not lose its transparent characteristic. A very thin layer may lead to reduced coverage with more recombination whereas a very thick layer may restrict the movement of electrons. Also, an optimal thickness of about 30-50 nm of SnO<sub>2</sub> ETL has been shown in studies to give the best performance in planar perovskite solar cells. Typically, an ALD cycle contains the following four steps: The first step involves exposing the first precursor to deposit

a film on the substrate within the reactor chamber; the second step involves the purging or removal of additional first precursor and byproducts; the third step involves exposing the second precursor; and the fourth step purge or evacuation of the excess second precursor and byproducts. Figure 3b depicts the deposition of ZnO via the ALD technique [37, 86]. ALD deposition is advantageous for flexible substrates since it allows the development of very crystalline layers at low growth temperatures. So far, ALD has shown to be an effective technique for producing low-temperature treated  $\text{MO}_x$  ETLs for high-grade PSC primarily including  $\text{TiO}_2$  [87],  $\text{ZnO}$  [88], and  $\text{SnO}_2$  [89] ETLs. However, large-scale production by the ALD technique might not be suitable due to its cost, duration, and reliance on the purity of the substrates.

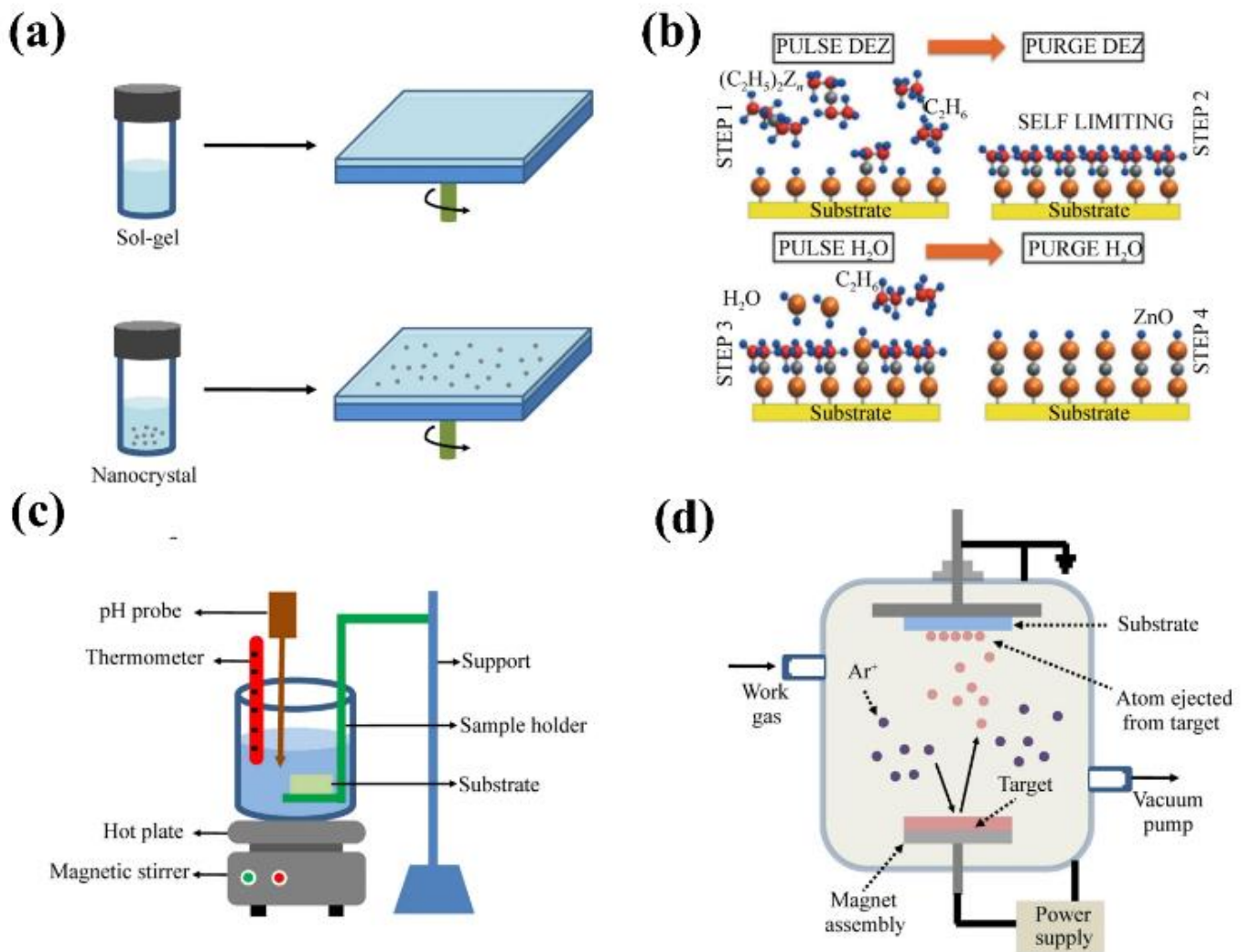
### Chemical Bath Deposition (CBD) Method

CBD is another well-established method for depositing semiconductor metal oxide layers by submerging the substrates in diluted solutions of metal ions as well as  $\text{OH}^-$  or

$\text{S}^{2-}$  ion sources as displayed in Figure 3c [90]. The CBD technique has proven to be ideal for the development of semiconductor layers particularly for  $\text{Cu}(\text{In, Ga})\text{Se}_2$  and  $\text{Cu}_2\text{ZnSnS}_4$  solar cells, like Cadmium sulfide, zinc oxide, or zinc sulfide [91, 92]. Particularly, recent studies have revealed that the CBD approach can be used to produce incredibly efficient  $\text{MO}_x$  ETLs for PSC such as titanium dioxide, zinc oxide, tin oxide, and iron oxide with definite arrangement [93, 94]. Low fabrication costs, simple growth adaptability, and excellent reproducibility are advantages of the CBD approach, which is beneficial in large-scale processing at low temperatures [38]. However, the temperature, concentration, and  $\text{pH}$  value during growth have a significant impact on the formation of thin layers.

### Other Deposition Methods

There are other efficient ways available to develop  $\text{MO}_x$  ETLs processed at low temperatures in accordance with various physicochemical features of substances in addition to the deposition processes discussed before.



**Figure 3.** Diagrammatic illustration of the common  $\text{MO}_x$  ETL deposition techniques: (a) solution process; (b) ALD for ZnO layer; (c) CBD; and (d) magnetron sputtering | adapted from Ref. [95].

For instance, electrochemical deposition (ED) is another significant technique for producing high-quality  $\text{MO}_x$  thin films and has previously been used to develop  $\text{TiO}_2$ ,  $\text{ZnO}$ , and  $\text{SnO}_2$  ETLs in conventional PSC [96-98]. The advantages of a facile, quick deposition and low-temperature preparation are exhibited by the ED approach. Additionally, by controlling the current density and deposition duration, it is possible to accurately regulate the thin film's properties after deposition with a firm connection to the substrate [50]. Other physical deposition methods used for the deposition of  $\text{MO}_x$  ETLs include magnetron sputtering displayed in Figure 3d and electron beam evaporation [99, 100]. These methods offer good potential for producing flexible electronics because  $\text{MO}_x$  can crystallize simultaneously with substrate deposition, effectively avoiding post-annealing. The drawbacks of electrochemical deposition and magnetron sputtering in depositing ETLs for PSCs include limited material choice, uniformity issues, and process complexity [101-103].

The main factors influencing the growth of  $\text{MO}_x$  ETL are substrate preparation, deposition parameters, and target material characteristics. It's also crucial to apply post-deposition treatments and get the appropriate film thickness. Moreover, charge transport and device stability can be strongly impacted by interface engineering at the intersections of ETLs and adjacent layers.

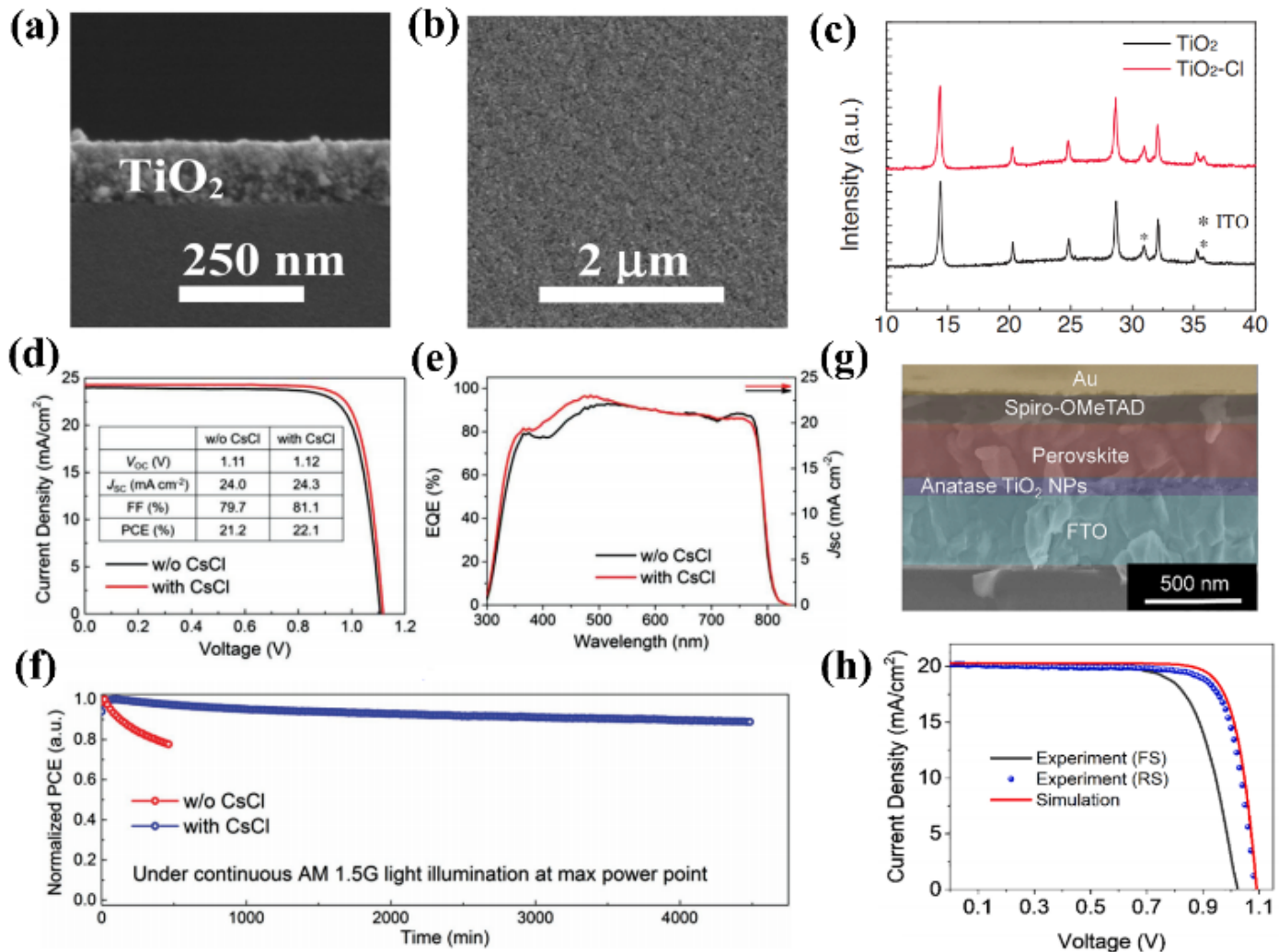
### **Different $\text{MO}_x$ used as an ETL Material in PSC**

#### **Titanium Oxide ( $\text{TiO}_2$ ) Based ETL in PSC**

$\text{TiO}_2$  thin layers annealed at low temperatures have been frequently created using the solution-processed approach.  $\text{TiO}_2$  nanoparticles (NPs) with good dispersion are typically favored for direct deposition on conductive substrates in planar PSC. An inverted PSC using a flat  $\text{TiO}_x$  film as ETL was first successfully fabricated by Snaith et al. [22]. The PSC treated at all low temperatures yielded a champion PCE of over 6% on flexible polymer substrates. Then, using a nonaqueous technique, they employed  $\text{TiCl}_4$  to synthesize  $\text{TiO}_2$  NPs. This resulted in a high-quality thin film of  $\text{TiO}_2$  devoid of cracks, as seen by the SEM picture in Figure 4a-b. [104]. The conductivity of the  $\text{TiO}_2$  film was greatly enhanced when titanium di-isopropoxide-bis was added, improving PCE up to 15.9% and proving the viability of processing at low temperatures in conventional PSC. Conings et al. proposed a low-temperature solution processing route for titanium dioxide ETL synthesized by facile one-pot fabrication NPs dispersion for regular PSC [105]. The  $\text{TiO}_2$  layer is also suitable for roll-to-roll fabrication on foils made of plastic since it just needs a heat treatment at 135 °C. To modify carrier behavior and boost conductivity, Zhou et al. deposited ETL at low temperature (150 °C) through yttrium (Y) doping into  $\text{TiO}_2$  nanocrystals [23]. The use of PEIE and Y-doped  $\text{TiO}_2$  lowered the Schottky limit at the ITO/ $\text{TiO}_2$  contact. The device performance was significantly increased to 19.3% due to improved transport and extraction of electrons.

Recently, Tan et al. observed that imperfect interfaces and charge recombination between perovskite and  $\text{TiO}_2$  interface may effect the PCE and stability of the PSC devices at low temperatures. As an alternative, they strengthened the  $\text{TiO}_2$ /perovskite interface by adding chlorine (Cl) to the  $\text{TiO}_2$  colloidal dispersion. This lowered the density of interfacial traps resulting in superior stability and a PCE of over 20 % via interface passivation. XRD patterns of perovskite layers on  $\text{TiO}_2$ -Cl and titanium dioxide are illustrated in Figure 4c [106]. As expected from the stoichiometric ratio of the precursors, XRD spectra show no obvious  $\text{PbI}_2$  nor other non-perovskite phases in films. It shows a smooth crystalline structure. Recently, Zhao et al. enhanced the quality of a perovskite thin layer by growing it over a low-temperature processed Cl-added  $\text{TiO}_2$  ETL. The successful deposition technique involving the addition of cesium chloride to the lead iodide solution resulted in the production of a high-quality perovskite layer. The champion device exhibited a superior PCE exceeding 22%, proving the effectiveness of  $\text{TiO}_2$  NPs as ETL and thus a viable option for PSC as in Figure 4d-f [107].

Several additional techniques were also employed to prepare  $\text{TiO}_2$  ETLs at low temperatures. An annealing-free titanium dioxide compact layer was originally employed using the ALD approach by Kim et al. for flexible planar PSC [85]. A 20 nm thick  $\text{TiO}_2$  compact layer was formed on the plastic conductive substrate less than 80 °C due to plasma accelerated ALD. They showed a very flexible PSC with a 1 mm bending radius, whereas 10 mm bending radius devices were able to sustain 1000 cycles while retaining 95% of their original PCE. Furthermore, Yella et al. tailored a nanocrystalline  $\text{TiO}_2$  layer at low temperatures via the CBD approach in PSC and achieved a reasonable PCE of 13.7 %. [109]. Due to the larger interfacial area compared to planar anatase  $\text{TiO}_2$ , the resultant rutile titanium dioxide was more efficient in extracting electrons from the perovskite layer. Mg-doped rutile  $\text{TiO}_2$  ETL treated at low temperature (70 °C) was also prepared by Liao et al. using the CBD approach, together with the carbon as an anode in PSC [110]. The optimal device exhibited a remarkable PCE of 15.73 %. Magnetron sputtering is a further intriguing method for titanium dioxide ETL deposition. Chen et al. deposited  $\text{TiO}_2$  ETL over a conductive substrate using low-temperature radio frequency (RF) magnetron sputtering. By using this technique, the thickness of the  $\text{TiO}_2$  layer could be modified to create a flexible PSC with a PCE of 8.9 % [111]. A planar flexible PSC sputtering process was further investigated by Yang et al [99]. A thick amorphous titanium dioxide ETL treated at low temperatures facilitated increased electron extraction from the layer of perovskite and enhanced electron movement, enabling the fabrication of a flexible device with a 15.07 % PCE. A recent optimization technique for RF magnetron sputtering encouraged the the enhancement in PCE of PSC to roughly 16% by utilizing flexible substrates with small, and uniform  $\text{TiO}_2$  nanoparticles-based ETL [112]. The device architecture must be optimized to achieve optimum performance in perovskite solar cells with  $\text{MO}_x$  ETLs.



**Figure 4.** (a) SEM cross-sectional image of glass coated with 3.54 (w/w%)  $\text{TiO}_2$  NPs with a thickness of 100–120 nm; (b) Top-view SEM image of low-temperature  $\text{TiO}_2$  [adapted from Ref. [104]; (c) XRD patterns of perovskite layer deposited atop  $\text{TiO}_2$  and  $\text{TiO}_2\text{-Cl}$  ETLs [adapted from Ref. [106]; (d) J–V characteristic curves are shown for the reverse scans (from 1.15 to  $-0.01$  V) with (red) and without CsCl (black); (e) EQE spectra of the same; (f) Testing the PSC's long-term stability without CsCl (red) and with CsCl (blue) at MPP under constant AM 1.5G illumination [adapted from Ref. [107]; (g) SEM cross-sectional image of PSC using the low temperature processed titanium dioxide ETL; and (h) Comparison of the J–V parameters for the top-performing planar PSCs between simulations and experiments (forward scan (FS) and reverse scan (RS)) [adapted from Ref. [108]

For example, the use of buffer layers, optimization of ETL thickness, and engineering an interface can vastly improve charge extraction and dramatically reduce recombination. This includes using a thin interfacial layer between the perovskite and MOX ETL to promote level alignment in energy and defect passivation, resulting in enhancements in PCEs. Furthermore, the devices exhibited steady power output (SPO) up to two hundred cycles of bending, showing the efficacy of the sputtering method. Shahiduzzaman et al. reported further development in treating  $\text{TiO}_2$  ETL at low temperatures to produce high-performing PSCs that are suitable for flexible substrates [113]. The optimized ETL was prepared by spin-coating single-phase crystalline anatase titanium dioxide nanoparticles with a median diameter of 6 to 10 nm. In comparison, the most efficient superstrate configuration of a planar device has a PCE of 17.1 % and a

$J_{sc}$  of  $20.3 \text{ mA/cm}^2$  as shown in Figure 4g-h. Lee et al. use a low-temperature processed bilayer of  $\text{SnO}_2/\text{TiO}_2$  as an ETL and achieve a PCE of 15.36% [114]. The deterioration of  $\text{SnO}_2$  can be stopped and the interfacial contacts between the light-absorbing layer and the ETL may be increased by using low-temperature-processed mesoporous  $\text{TiO}_2$ . The power conversion efficiency of  $\text{SnO}_2/\text{TiO}_2$  bilayer-based PSCs was much greater than that of single  $\text{SnO}_2$  ETL-based PSCs. Nam et al. produced mesoporous-structured  $\text{TiO}_2$ -based PSCs with outstanding photovoltaic performance at low temperatures for the first time by demonstrating a successive surface engineering technique using oxygen plasma treatment [115]. The low-temperature solution-processed  $\text{TiO}_2$  layer's organic additives were effectively eliminated by the oxygen plasma treatment. Increases in  $\text{MAPbI}_3$  particle size and infiltration depth into the mesoporous-structured  $\text{TiO}_2$  layer were brought

about by the oxygen plasma treatment's enhanced wettability of TiO<sub>2</sub> layers. Furthermore, the PCE of 14.82% was attained by reducing the oxygen vacancies in the TiO<sub>2</sub> layer, which served as trap sites, using an oxygen plasma treatment.

### Zinc Oxide (ZnO) Based ETL in PSC

ZnO NPs solutions offer certain unique characteristics; such as a straightforward and controllable synthesis route and the ability to fabricate thin layers processed at low temperatures. These unique features have been thoroughly investigated for the utilization of ZnO as ETL in PSC. Liu and Kelly initially introduced ZnO NPs-based ETL in PSC in 2014 [32]. The compact pinhole-free ETL of ZnO NPs was prepared by employing a solution-processed approach by mixing KOH and Zn(CH<sub>3</sub>COO)<sub>2</sub>·2H<sub>2</sub>O dispersions and further low-temperature post-annealing treatment. The high crystallinity of ZnO NPs exterminated the need for high-temperature post-treatment. The perovskite layer deposited atop ZnO ETL exhibited large grain sizes with visible grain boundaries and the corresponding devices showed PCE of 10.2 % and 15.7 %, for flexible and rigid substrates, respectively.

Hwang et al. stated a complete slot die-coated PSC assembled of ZnO NPs ETL using a handmade 3D printer under ambient conditions. [116]. The optimized PCE of 11.96 % provided the way forward for low-cost and bulk manufacturing of PSC. In the meanwhile, a ZnO/perovskite/carbon planar structure PSC with a metal-electrode-free design and HTL was reported by Zhou et al. Moreover, flexible devices' mechanical strength can be greatly increased by employing this special structure [117]. Song et al. applied industrial ZnO NPs to produce ETL at low temperatures for PSC using a spin coating technique and used di-amino methyl carbonium ion instead of methylamine to produce the absorber material by consecutive layering approach to overcome the problem of thermal instability in ZnO-based PSC as illustrated in Figure 5a-d [118].

The perovskite layer's thermal stability deposited atop zinc oxide ETL was significantly increased with formamidinium lead iodide due to the fundamental resilient character of formamidinium contrasted to methylammonium in methylammonium lead iodide, and the ZnO-FAPbI<sub>3</sub> planar structures exhibited PCE of 16.1% after optimization. FAPbI<sub>3</sub> was further substituted with triple-cation perovskite, the PSC achieved a higher PCE of 18.9 % with superior environmental photostability and durability [119]. Additionally, when FAPbI<sub>3</sub> was replaced with triple cation perovskite, the same ZnO NPs-based ETL might be used to develop a steady absorber of light using a single-stage deposition technique. A highly transparent ZnO NPs solution was prepared recently using an ultrasonic-assisted approach, which made it possible to prepare a denser and more homogeneous ZnO layer. In addition, a room-temperature aging stage of ZnO thin film was also investigated in this work to enhance the thermal resistance between ZnO and MAPbI<sub>3</sub> interface. High-quality ZnO and perovskite thin films contributed synergistically to prevent device degradation, showing good stability even after

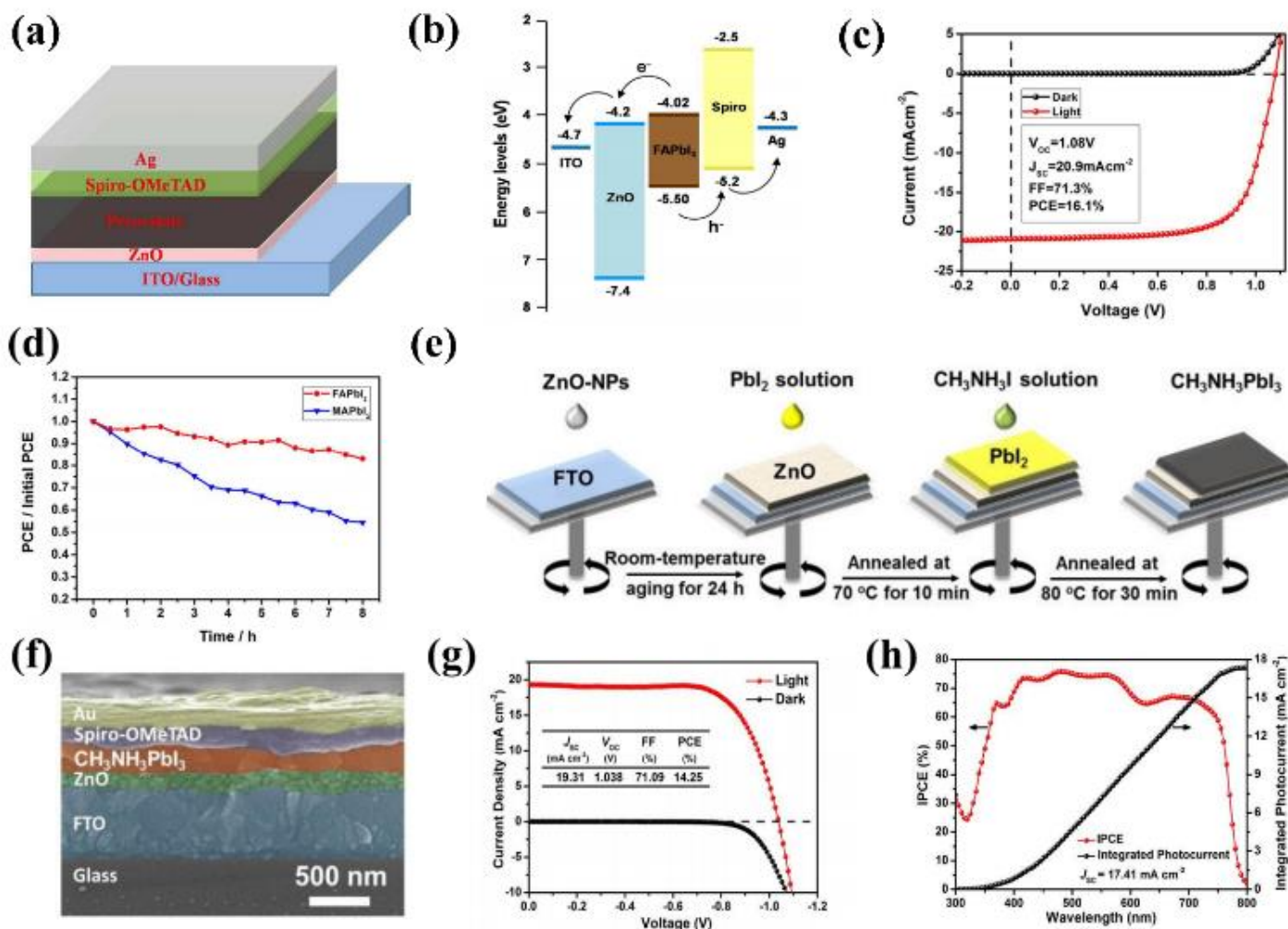
being exposed to air after 45 days. The SEM images and JV curves are shown in Figure 5e-h [120].

Yang et al. described a method to synthesize ZnO ETL at low temperatures and treated it with an aqueous solution approach employing [Zn(NH<sub>3</sub>)<sub>x</sub>](OH)<sub>2</sub>, a solution containing an ammine-hydroxo zinc combination [121]. Consequently, a device with an enhanced V<sub>OC</sub> of 1.07 V was developed by lowering the ZnO layer annealing temperature to 150 °C. The ZnO NPs atop the perovskite layer allowed inverted PSC to acquire both inorganic charge transport layers (ETL and HTL) and significantly increased device stability with a stabilized PCE of 16.1 % [118]. Similar to this, Savva et al. used aluminum-doped ZnO (AZO) ETL in an inverted PSC [122]. The characteristics of AZO as MO<sub>x</sub> ETL can be further optimized by adjusting energy levels to recognize charge carrier transport, and eliminating the pinholes resulting in an enhanced PCE, improved durability, and stability of PSC.

In a quest to find alternative deposition techniques, Lee et al. used the ALD technique to prepare compact ZnO ETL at 80 °C for planar PSC following the comprehensive analysis of film thickness and morphology [123]. Yuan et al. achieved a PCE of 13.1% using this technique [88]. They revealed that the precursor utilized for the formation of a compact ZnO layer using ALD kept at ambient temperature might assist in the formation of CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub>. To further develop low-temperature and solution-processed, flexible PSCs, Mathews et al. utilized ZnO nanorods to develop a compact ETL for PSC using the CBD and ED techniques [94]. Zhang and Pauportè used the ED technique to deposit ZnO ETL processed at low temperatures [97]. They observed and compared how the "one-step" and "two-step" techniques for making CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> were affected by the ZnO and TiO<sub>2</sub> ETLs. The outcomes showed that an ideal device efficiency of 15% might be obtained via combining a perovskite film made by a "one-step" method with a little overvoltage electrodeposited ZnO layer. The impact of RF magnetron sputtering settings on the formation of zinc oxide ETL quality was investigated by Tseng et al. [124]. They discovered that the electrical and electronic characteristics of zinc oxide film were significantly influenced by O<sub>2</sub> voids, it could be managed by altering the ratio of argon to oxygen used as working gases during sputtering. Sputtering was found to be the optimal way to develop ZnO ETLs with reasonable characteristics, and a PCE of 15.9 % was achieved under the Ar environment. Additionally, they used an analogous approach to develop a high-quality film of Al-doped ZnO on an ITO substrate with complete coverage. This layer showed improved band alignment with methylammonium lead iodide, better conductivity, and superior acid resistance than zinc oxide [125].

Consequently, the performance of the PSC using the modified ETL was significantly enhanced. Lai et al. used a C-60 interlayer acting as a protective layer between the sputtered ZnO and perovskite to prevent sputtering damage to the absorber layer and contributed to better device performance [126]. Similarly, Liu et al. modified the CH<sub>3</sub>NH<sub>3</sub>Cl to increase the stability of ZnO-based PSC [127].





**Figure 5.** (a) Schematic diagram of ITO/ZnO/FAPbI<sub>3</sub>/spiro-OMeTAD/Ag PSC; (b) Energy level diagram of the same; (c) Light (red line) and dark (black line) J-V curves (black line); (d) Photostability of FAPbI<sub>3</sub> and MAPbI<sub>3</sub> absorber material based PSC [adapted from Ref. [118]]; (e) Diagrammatic representation of the steps involved in preparing a stable ZnO/MAPbI<sub>3</sub> interface by combining a two-stage sequential spin coating technique for the methylammonium lead iodide layer with an aging step for the ZnO layer; (f) SEM cross-sectional image of a complete PSC using zinc oxide ETL; (g) For the ideal zinc oxide based PSC device, the JV curve was evaluated in the dark (black line) and under 100 mW cm<sup>-2</sup> AM 1.5G illumination (red line); and (h) EQE spectrum of the same [adapted from Ref. [120]].

The PSC was fabricated with the configuration ITO/ZnO–MACl/MAPbI<sub>3</sub>/Spiro-OMeTAD/MoO<sub>3</sub>/Ag. The MACl-modified devices exhibited superior photovoltaic parameters (J<sub>sc</sub> of 21.96 mA/cm<sup>2</sup>, V<sub>oc</sub> of 1.10 V, FF of 77.1 %, and PCE of 18.7%) relative to unmodified pristine ZnO-based devices (PCE of 16.55 %).

### Tin Oxide (SnO<sub>2</sub>) Based ETL in PSC

SnO<sub>2</sub>, a naturally occurring, nontoxic, inherently n-type semiconductor with unique characteristics such as excellent stability, superior electron mobility, and high transparency, holds enormous potential as an ETL material in PSC. Tin-based precursors like tin chloride, tin tetrachloride, or hydrates like tin chloride dihydrate and tin chloride pentahydrate are frequently utilized to produce tin oxide thin layers in the sol-gel method. Ke et al. used a precursor of

SnCl<sub>2</sub>·2H<sub>2</sub>O to develop SnO<sub>2</sub> thin films through thermal annealing in ambient for one hour at 180 °C in 2015 [82]. After a 15-minute UV-ozone treatment, it became feasible to deposit the perovskite layer atop tin oxide ETL and the champion device showed a PCE of 17.21 %. Zuo et al. used a similar technique for creating a SnO<sub>2</sub> ETL-based perovskite solar cell with an exceptional PCE of 20.23% after upgrading perovskite with a polymer template as displayed in Figure 6a-c [128]. Dong et al. suggested an alternate sol-gel technique by refluxing atmospheric oxygen and water (reflux condensation technique), which considerably accelerated the decomposition and oxidation of SnCl<sub>2</sub>·2H<sub>2</sub>O precursor in an ethanol-based solution and reduced the thermal decomposition temperature as shown in Figure 6d. SnO<sub>2</sub> precursor solution might be cooled below 80 °C as a result.

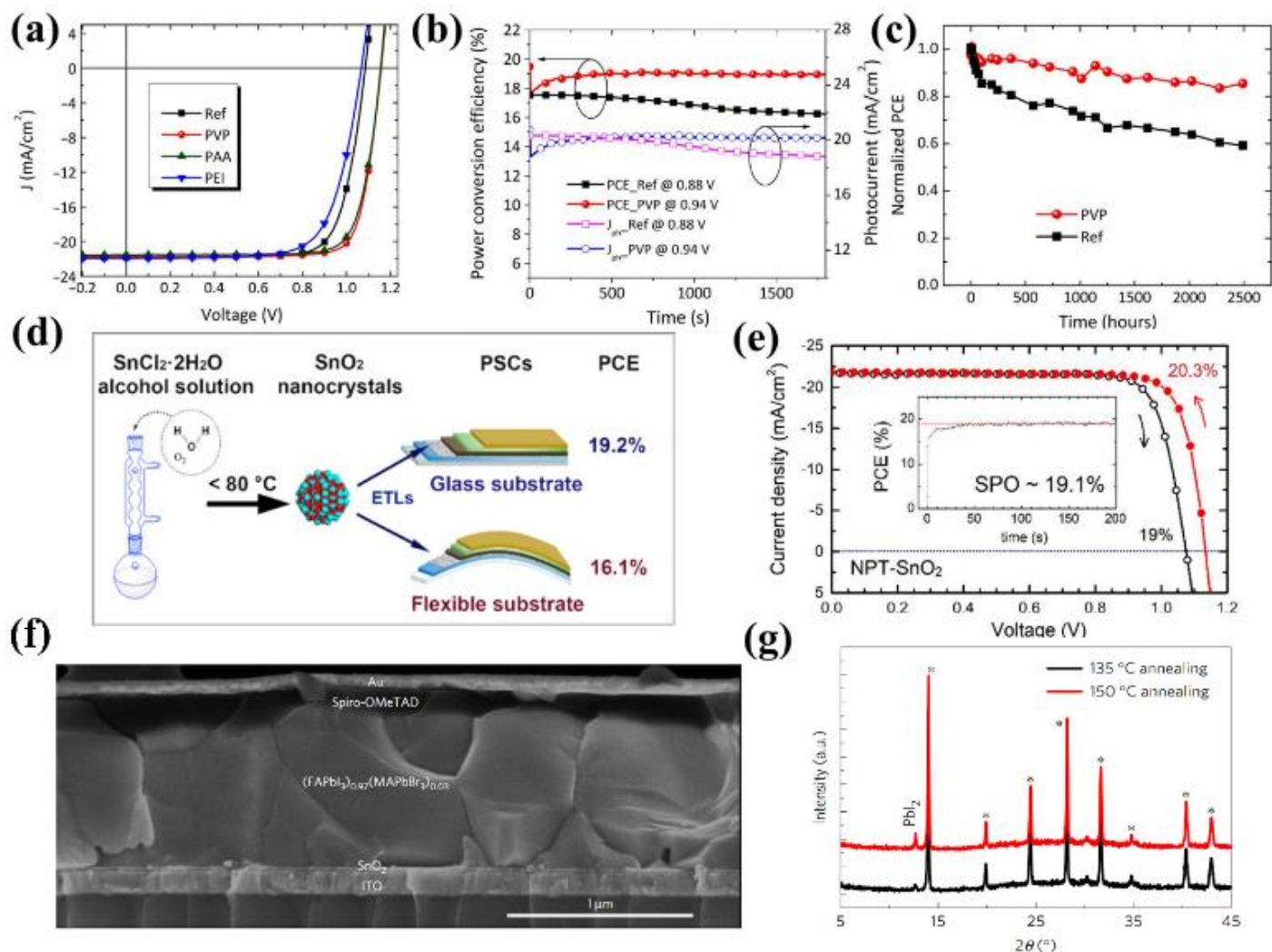
On both flexible and stiff substrates, the regular planar PSC devices utilized this SnO<sub>2</sub> as ETL and achieved PCE of 19.2% and 16.11%, respectively [129]. Subbiah et al. activated the precursor layer at ambient temperature using a low-power nitrogen RF Plasma. The strong ultraviolet photons from the nitrogen plasma broke -OR and -OH groups, which aided in the creation of the MO<sub>x</sub> metal framework. The PSC relying on nitrogen plasma processed tin oxide ETL manifested a PCE of 20.3 % on stiff substrates. On PET substrates, a flexible PSC was also fabricated using the aforementioned technique. The PSC exhibited an outstanding PCE of 18.1 % by retaining 90 % of its original PCE even after a thousand hours as illustrated in Figure 6e [130]. In contrast, the high-temperature annealing is avoidable since the SnO<sub>2</sub> NPs are prepared prior to deposition. Song et al. dispersed industrial-grade tin oxide nano particles (approximately 22 to 43 nm) in C<sub>4</sub>H<sub>9</sub>OH to develop a thin compact ETL via spin-coating at temperatures in PSC [81]. However, the SnO<sub>2</sub> NPs exhibited irregular morphology with random size distribution of nanoparticles resulting in a low device efficiency of 13 %.

Additionally, You et al. utilized commercial tin oxide NPs colloidal solution to synthesize a pinhole-free compact ETL with a uniform particle size of 3 to 4 nm processed at low temperatures as shown in Figure 6f,g [131]. Using a PbI<sub>2</sub> passivation, a certified efficiency of device 19.9 % was attained in PSC. By further adjusting lead iodide contents, perovskite solar cells subsequently showed efficiencies of 21.6 % for small-size devices (0.0737 cm<sup>2</sup>) and 20.1 % in large sizes (1 cm<sup>2</sup>), with a certified PCE of 20.9 % for small-size devices [132]. In 2019, the same group reported less non-radiative recombination in PSC by reducing the interfacial defects, and higher efficiency PSC was fabricated due to the passivation of surface imperfections/defects using PEAI. The optimal devices exhibited a higher V<sub>OC</sub> of 1.18 V with a certified PCE of 23.32 % [133]. A slot die approach for printing good quality tin oxide layers for effective flexible PSC was developed by Bu et al. to examine the solution printability of tin oxide nano particles on plastic substrates [134]. The interfacial contacts were passivated using a general potassium passivation approach to reduce the inherent hysteresis developed by SnO<sub>2</sub> ETL-based PSC. The performance of the small-size flexible PSC was 17.18 %, and the efficiency of the large-size (5 cm to 6cm) flexible modules was over 15 % with no hysteresis.

In addition to commercial SnO<sub>2</sub> colloidal NPs, Yang et al. synthesized SnO<sub>2</sub> quantum dots by mixing thiourea and SnCl<sub>2</sub>·2H<sub>2</sub>O to produce excellent quality tin oxide quantum dots with variable carrier concentrations. A highly stable PCE of 20.32 % with FTO glass substrates and 16.97 % on flexible PEN substrates was obtained for conventional PSC [135]. Similar to TiO<sub>2</sub> and ZnO, SnO<sub>2</sub> thin films can be prepared via the ALD technique. Baena et al. employed ALD to develop compact tin oxide thin layers, which displayed an optimistic and balanced energy level with the perovskite layer [136]. The PSC demonstrated a behavior free of hysteresis with a remarkable PCE of 18 % and a very high V<sub>OC</sub> of 1.19 V. Wang et al. used plasma-enhanced atomic layer deposition (PE-ALD) to reduce the temperature of deposition (<100°C) to

enable low-temperature industrial production of PSC [137]. On flexible substrates and glass, tin oxide ETL-based devices showed PCE of 19.03 % and 16.80 %, respectively, revealing the potential of this low-temperature deposition method. Additionally, they induced the formation of the SnO<sub>2</sub> layer by introducing water vapor, potentially facilitating a more comprehensive reaction of the organic precursor and resulting in the production of highly purified tin oxide. A notable PCE of 18.36% was achieved by the flexible PSC based on the obtained SnO<sub>2</sub> [138]. SnO<sub>2</sub> films were also prepared using the CBD technique. A straightforward approach combining SC and CBD to produce SnO<sub>2</sub> ETL was disclosed by Anaraki et al. [139]. They spin-coated a SnO<sub>2</sub> seed layer first, and as a post-treatment on the spin-coated films, they grew a SnO<sub>2</sub> layer using the CBD approach. The SC-CBD approach for depositing the SnO<sub>2</sub> layer showed effective hole-blocking abilities and increased repeatability. The resulting PSC has a high V<sub>OC</sub> of 1.214 V and an efficiency of 20.7 % under maximum power point tracking. Bu et al. further used the CBD approach [140]. A small-size perovskite solar cell (6x6 cm) sub-module was combined with a unique quadruplication perovskite to produce PCEs of 20.56 % and 15.76 %, respectively.

In addition, several new deposition methods have emerged, including electrochemical deposition [98], combustion [141], pulsed laser deposition [142], and E-beam evaporation [100]. E-beam evaporation stands out among them as a viable option since it allows for the simultaneous production of hundreds of SnO<sub>2</sub> ETL substrates, which justifies industrial use. Fang et al. demonstrated the effective SnO<sub>2</sub> ETL-based planar perovskite solar cell using this deposition method with the highest PCE of 18.2 % without any interface modification [100]. Kim et al. examined the impact of K treatment on the tin oxide electron transport layer. It was noticed that the use of KCl treatment boosted the CBM at the perovskite/SnO<sub>2</sub> interface by raising the Fermi level and conduction band maximum energy of tin oxide and narrowing the band gap of the perovskite absorber through K<sup>+</sup> diffusion. J-V hysteresis was nearly diminished due to improvements in the alignment of the band and passivation of defects achieved by the potassium chloride treatment [143]. Huang et al. reported on the low-temperature UV curing procedure used to make SnO<sub>2</sub> films and their use as the electron-transporting layer in planar PSCs [144]. The formation of these SnO<sub>2</sub> ETLs involves subjecting the as-deposited precursor films to UV irradiation at a very low temperature (70 °C) without any additional thermal or chemical treatment. Using these films acting as ETLs, photovoltaic devices demonstrated the greatest efficiency of 16.21%, significantly greater than those using SnO<sub>2</sub> ETLs made using a traditional thermal annealing method (11.49%). Shekargoftar et al. investigated the preparation of SnO<sub>2</sub> using plasma treatment at a low temperature of 60 °C [127]. A set of SnO<sub>2</sub> films treated by traditional thermal annealing was contrasted with SnO<sub>2</sub> films treated by plasma. Films with homogenous surfaces and a highly crystalline structure were produced by plasma treatment.



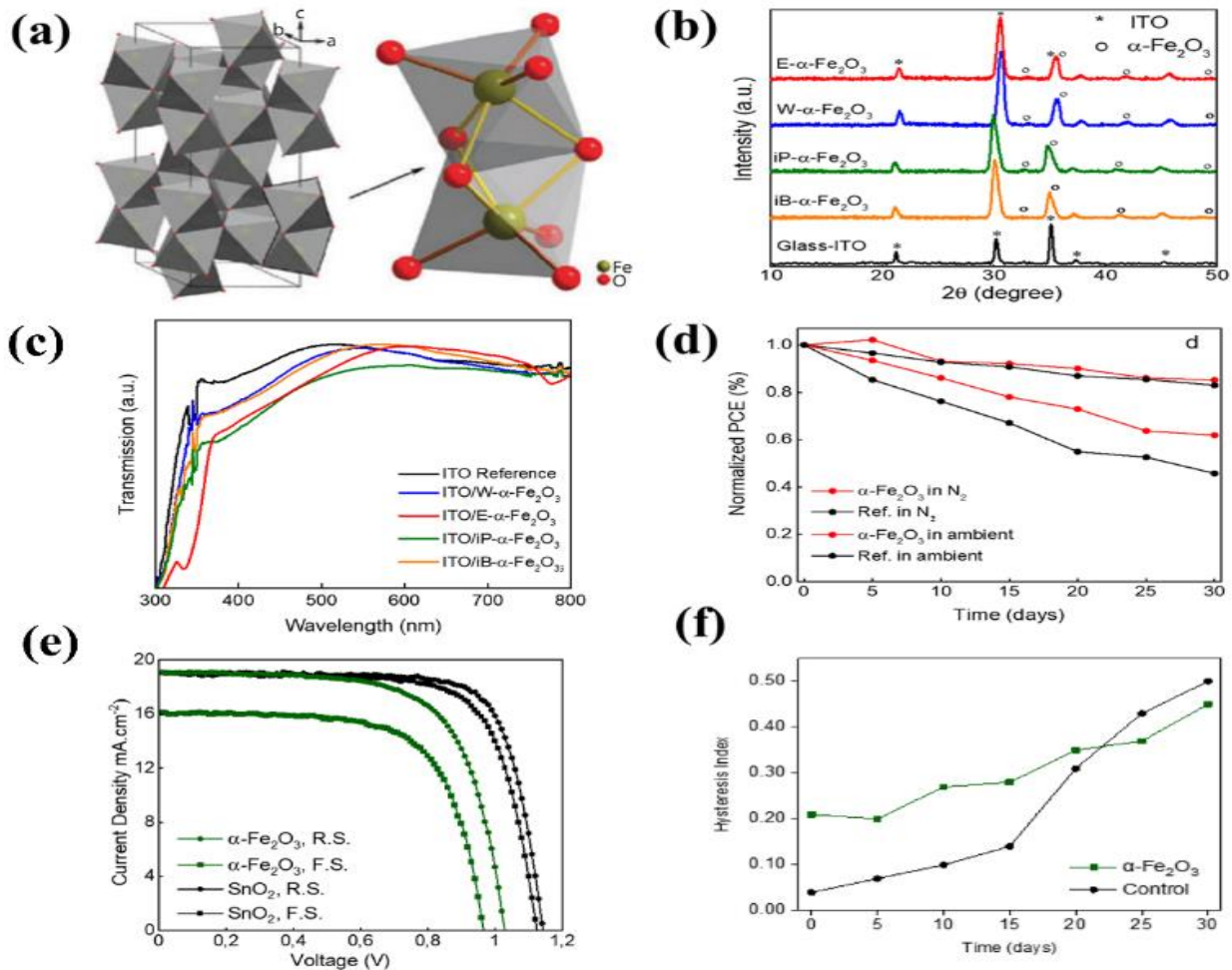
**Figure 6.** (a) J-V curves; (b) EQE spectra; (c) shelf-life stability with and without PVP treatment of  $\text{CH}_3\text{NH}_3\text{PbI}_3$  solar cell devices with various polymers converted from microporous  $\text{PbI}_2$  [adapted from Ref. [128]]; (d) Novel approach to preparing low-temperature  $\text{SnO}_2$  nanocrystals [adapted from Ref. [129]]; (e) Planar PSC with light JV curves at  $100 \text{ mW/cm}^2$  (1 sun) intensity using TA- $\text{SnO}_2$  and NPT  $\text{SnO}_2$  layers as ETL in both FS and RS orientations produced a PCE of 20.3% [adapted from Ref. [130]]; (f) SEM cross-sectional image of PSC with configuration glass/ITO/ $\text{SnO}_2$ /(FAPbI<sub>3</sub>)<sub>0.97</sub>(MAPbBr<sub>3</sub>)<sub>0.03</sub>/Spiro-OMeTAD/Au; (g) XRD patterns of perovskite films annealed at 130 °C and 150 °C, respectively [adapted from Ref. [131]].

Only very few variations in the charge extraction of the  $\text{SnO}_2$  layers produced by heat annealing and plasma were found upon further investigation. The qualities of films created by plasma deposition are quite competitive when compared to films created using thermal annealing. Their findings revealed that the efficiency of this plasma treatment reached 15.17%, a value comparable to the efficiency achieved through thermal annealing, which stood at 15.91% for  $\text{SnO}_2$ .

### Iron Oxide ( $\alpha\text{-Fe}_2\text{O}_3$ ) Based ETL in PSC

Hematite ( $\alpha\text{-Fe}_2\text{O}_3$ ) is a thermodynamically stable n-type semiconducting iron oxide that has attracted attention due to its low UV photocatalytic activity, which is useful for prolonging the UV stability of PSCs [53]. The crystal structure and light absorption profile of hematite are shown in

Figure 7a.  $\alpha\text{-Fe}_2\text{O}_3$  has an optical band gap of 2-2.3 eV and is a cheap, abundant, and non-toxic material [145-147]. Since  $\alpha\text{-Fe}_2\text{O}_3$  has a lower conduction band minimum (CBM) than  $\text{TiO}_2$ , it facilitates the process of extracting electrons from the perovskite layer [148]. Hu et al. utilized  $\alpha\text{-Fe}_2\text{O}_3$  as the charge transfer layer (CTL) in  $\text{MAPbI}_3$  PSC, exhibiting 11 % PCE and enhanced stability in comparison to a  $\text{TiO}_2$ -based PSC. In the  $\alpha\text{-Fe}_2\text{O}_3$  ETL-based PSC, the significant potential across the perovskite film promoted better extraction of charges and less accumulation of charges at the interface [66, 149]. Guo et al. utilized nickel-doped  $\alpha\text{-Fe}_2\text{O}_3$  ETL in planar heterojunction PSC. The dopant addition might increase electronic conductivity and cause a reduction in the injection of electrons and transfer from the perovskite conduction band..



**Figure 7.** (a) Crystal Structure of  $\alpha\text{-Fe}_2\text{O}_3$ ; (b) XRD patterns  $\alpha\text{-Fe}_2\text{O}_3$  films deposited atop ITO using different solvents (deionized water, ethanol, isopropanol, and isobutanol) [adapted from Ref. [110]; (c) Transmission spectra of ITO/ $\alpha\text{-Fe}_2\text{O}_3$  thin layers with various solvents; (d) shelf-life stability conducted using identical conditions for the control and champion devices [adapted from Ref. [153]; (e) J-V curves at 10 mg/mL concentration of hematite EEL and a  $\text{SnO}_2$  EEL control device; and (f) Champion and control device hysteresis index [adapted from Ref. [154].

Due to this, the device became substantially less responsive to direction and rate of scanning, or reduced hysteresis, due to a high reduction in the charge buildup at the perovskite/ETL interface. The optimized PSC shows a 14.2 % competitive PCE with a high  $J_{sc}$  of  $22.35 \text{ mA/cm}^2$ ,  $V_{oc}$  of  $0.92 \text{ V}$ , and FF of 69.1% when scanned in reverse under standard AM-1.5 sunlight illumination. In addition, it is possible to obtain good stability for devices exposed to high UV radiation levels and ambient air [150]. Hou et al. used an  $\alpha\text{-Fe}_2\text{O}_3$  fullerene bilayer in a planar PSC and demonstrated a PCE of 14 % (2-38.) With a low trap density and strong electron mobility, Zhu et al. were able to construct  $\text{Ti-Fe}_2\text{O}_3$  ETLs with a PCE of 17.8 % and better charge transfer [151]. Guo and colleagues utilized interfacial engineering to develop a bifunctional alteration of the  $\alpha\text{-Fe}_2\text{O}_3$ /perovskite interface through the use of  $\text{PbI}_2$ . This led to a notable enhancement in the alignment of energy levels and the suppression of JV hysteresis [152]

Qureshi et al. examined the impacts of solvents on the optoelectronic characteristics of  $\alpha\text{-Fe}_2\text{O}_3$  thin films by working on the solvent-assisted crystallization of ETLs. An

n-i-p-configured PSC with an optimized ethanol-based (0.2M)  $\alpha\text{-Fe}_2\text{O}_3$  ETL demonstrated a decreased hysteresis index of 0.04 and a PCE of 13%, out of the several solvents utilized in their investigation (de-ionized water,  $\text{C}_2\text{H}_5\text{OH}$ , isopropanol, and iso-butanol). They reported that the development of a compact shape of ETLs without pinholes leads to improved charge transfer efficiency, decreased interfacial recombination, and free-of-crack surface coverage of the perovskite layer atop an  $\alpha\text{-Fe}_2\text{O}_3$  ETL as shown in Figure 7b-d [153]. To prepare low-temperature solution-processed hematite ETLs, Qureshi et al. used natural  $\alpha\text{-Fe}_2\text{O}_3$  as an ETL. The completed device manufacturing was done at a temperature lower than  $150 \text{ }^\circ\text{C}$  to demonstrate the cost-effectiveness of this innovative electron extraction layer (EEL). The optimal concentration-based device showed a PCE of 13.3 % and a  $V_{oc}$  of  $1.03 \text{ V}$  with long-term shelf life stability. The enhanced photovoltaic properties are due to the smooth surface structure of CsFAMA atop natural  $\alpha\text{-Fe}_2\text{O}_3$  EEL, high crystallinity of natural  $\alpha\text{-Fe}_2\text{O}_3$ , and lower interfacial recombination as in Figure 7e-f [154].

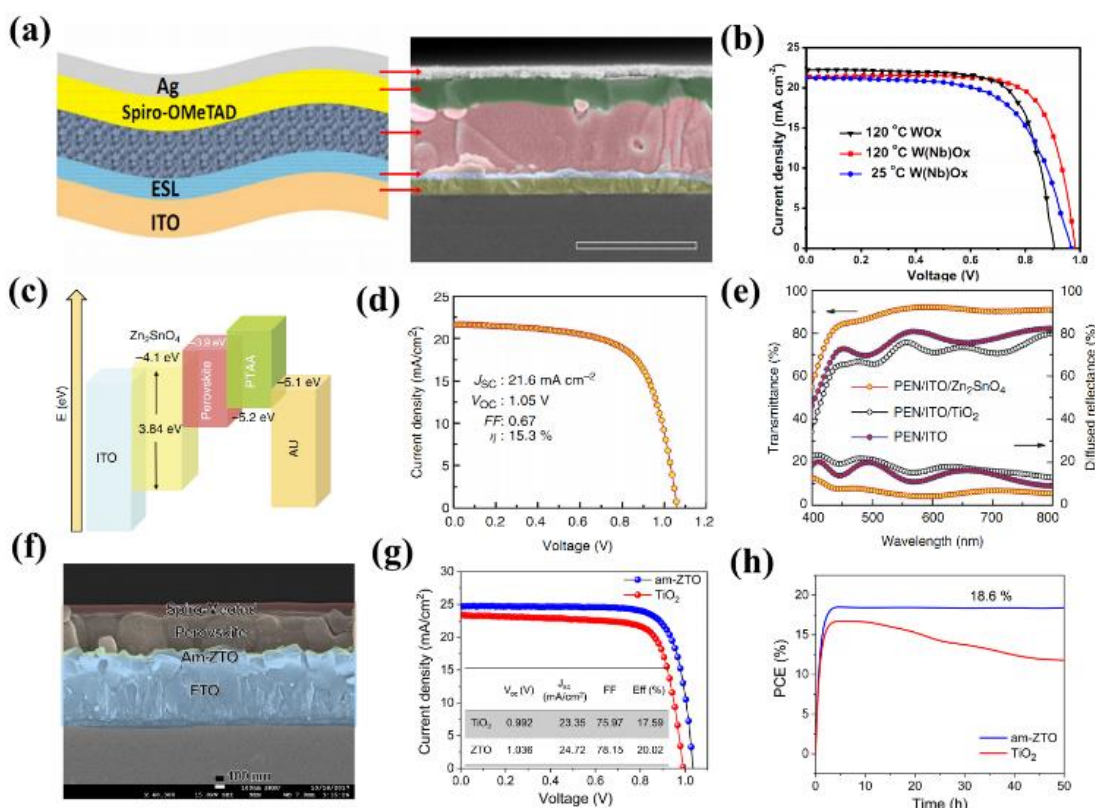
## Other Metal Oxide ETL Based in PSC

It is possible to look at other binary  $\text{MO}_x$  that are less renowned as compared to  $\text{TiO}_2$ ,  $\text{SnO}_2$ ,  $\text{ZnO}$ , and  $\alpha\text{-Fe}_2\text{O}_3$  ETL materials. Dong et al. prepared a chromium oxide ( $\text{Cr}_2\text{O}_3$ ) layer on an FTO glass substrate with SC chromium oxide ink without post-annealing [61]. The chromium oxide layer demonstrated great optical transmittance and effective electron transport. A PCE of 16.23% was attained by the planar PSC based on  $\text{Cr}_2\text{O}_3$  ETL. Similarly, Wang et al. produced cerium oxide ( $\text{CeO}_x$ ) layers at 150 °C employing a basic sol-gel technique [64]. The PCE in PSC was raised to 17.04 % after  $\text{CeO}_x$  was modified with PCBM, and it displayed greater stability under illumination. Hu et al. employed a solution-processing technique at a low temperature of 100 °C to develop  $\text{CeO}_x$  ETLs in inverted configuration PSC [155]. The PSC showed an improved PCE of 17.1 % with long-term inert and ambient stability.

Tungsten Oxide ( $\text{WO}_3$ ) is another novel ETL material recently used in PSC with an efficient hole-blocking capability. Wang et al. utilized niobium-modified tungsten oxides ETL modified with plastic substrates at low-temperature [156]. The addition of  $\text{Nb}^{5+}$  significantly enhanced the PCE up to 15.65 % as shown in Figure 8 a-b. Without any post-treatment, Feng et al. developed  $\text{Nb}_2\text{O}_5$  ETL

by the E-beam evaporated method [59]. Champion PSC devices using rigid and flexible substrates showed PCE values of 18.59 % and 15.56 %, respectively at an optimized thickness of 60-nm  $\text{Nb}_2\text{O}_5$  layer. Furthermore, the E-beam evaporated niobium oxide is advantageous for flexible PSC in wide areas, and corresponding  $J_{\text{SC}}$  and  $V_{\text{OC}}$  values are comparable to those of small-area devices. The primary cause of losses in PCE, which led to a reduced FF, was increased series resistance.

Ternary  $\text{MO}_x$  differs from binary  $\text{MO}_x$  owing to their exceptional characteristics qualities including high crystallization and complete surface coverage. Furthermore, the physical & chemical characteristics of ternary  $\text{MO}_x$  can be enhanced by modifying and tuning their compositions. Shin et al. employed hydrazine to produce  $\text{Zn}_2\text{SnO}_4$  at a low processing temperature [157]. A PCE of 15.3 % was observed for ternary  $\text{MO}_x$  ETL-based flexible PSC as shown in Figure 8c-e. Using zinc-stannate nano particles and quantum dots, they also customized the energy levels in the modified electron transport layer. The energy level graded electron transport layer produced by the sequential deposition of nano particles & quantum dots on a flexible substrate demonstrated enhanced electron collection efficiency and boosted power conversion efficiency to 16.5 %. [158].



**Figure 8.** (a) Cross-sectional SEM image displaying each functional layer in the flexible PSC; (b) J-V curves with multiple ESL [adapted from Ref. [156]; (c) PSC energy levels; (d) Under AM 1.5 G illumination, the photocurrent (JV) curve was measured by reverse scanning with 10 mV voltage increment and 40 ms delay intervals; (e) PEN/ITO/ZSO, PEN/ITO/ $\text{TiO}_2$ , and PEN/ITO substrate transmittance and reflectance spectra [adapted from Ref. [157]; (f) SEM cross-sectional image of PSC with Am-ZTO ETL; (g) JV curves of PSC using am-ZTO ETL and  $\text{TiO}_2$  as reference ETL; and (h) MPPT of an Am-ZTO ETL based PSC [adapted from Ref. [159].

Using a sol-gel technique, Jung et al. prepared amorphous  $\text{Zn}_2\text{SnO}_4$  layers with superior surface homogeneity, fast mobility of electrons, and low-voltage trapping [159]. The resulting perovskite solar cell had better device stability, and negligible hysteresis behavior, and produced a PCE of 20.02 % as shown in Figure 8f-h.  $\text{BaSnO}_3$  is another semiconducting ternary metal oxide with a large band gap of 3.2 eV. However, large temperature treating of  $\text{BaSnO}_3$  around 900 °C severely limits its use in flexible devices [160]. Shin et al. produced a colloidal solution of superoxide molecular clusters for the deposition of La-doped  $\text{BaSnO}_3$  films at temperatures lower than 300 °C [161]. The devices utilizing  $\text{BaSnO}_3$  ETL in PSC showed superior PCE of 21.2 % and remained at 93% of its initial PCE up to one thousand hours. However, 300 °C is still not a suitable temperature for plastic substrates. To avoid this, Sun et al. manufactured completely dispersed barium stannate nanoparticles via the facile peroxide-precipitate technique and employed them as an ETL in n-i-p configured PSC with low temperature (150 °C) post-treatment. The PSC exhibited a comparative PCE of 10.96 % [72].

The deficiencies of binary  $\text{MO}_x$  as ETL material were further alleviated by mixing two semiconductors with compensating properties [162]. In comparison to pure  $\text{WO}_x$ , Wang et al. observed that adding  $\text{TiO}_x$  to  $\text{WO}_x$  would increase the Fermi level and reduce interfacial recombination of charge [163]. Consequently, by optimizing the  $\text{MO}_x$  molar ratios, the PSC achieved a reasonable PCE of 14.47 %. Similarly, Zinc oxide-tin oxide nanocomposite-based thin layers were prepared by Song et al. by simply combining  $\text{ZnO}$  and  $\text{SnO}_2$  NPs [164]. The devices exhibited photovoltaic performance dependent on  $\text{ZnO}/\text{SnO}_2$  ratio. The planar PSC demonstrated a comparatively high efficiency of 14.3 % with an optimum weight ratio of 2:1. It was noted that the devices with  $\text{ZnO}-\text{SnO}_2$  ETL realized significant thermal stability compared to bare  $\text{ZnO}$  ETL-based PSC.

## Conclusion and Outlook

The low-temperature solution-treated  $\text{MO}_x$  are potential applicants as ETLs for effective planar perovskite solar cells, which can drastically decrease the cost of fabrication and be advantageous for flexible and large-module solar cells. In order to comprehend the connection between  $\text{MO}_x$  ETL and device performance, this review gives an extensive summary of the fundamental characteristics of different  $\text{MO}_x$  materials, including energy levels, processability, mobility, transparency, and even stability. Then, a brief description of commonly used methods for depositing  $\text{MO}_x$  ETL was discussed. Finally, a detailed discussion and demonstration of the most recently developed  $\text{MO}_x$  ETLs in planar PSC with superior performance produced at low temperatures are provided. Single-step processed  $\text{MO}_x$  ETLs at low-temperatures also have some limitations despite the advantages listed above. Uniform film quality and their crystallinity at low temperatures are not easily obtained. The presence of interfacial defects and the possibility for chemical incompatibility with the perovskite layer could be responsible to some extent for poor device performance. For addressing these challenges, surface modifications and doping strategies

play a vital role in further improvement. The following critical requirements must be met by an ideal  $\text{MO}_x$  ETL for best-resulted PSC: (1) outstanding film-forming capabilities via facile and low-cost deposition methods; (2) appropriate energy levels to enable successful charge extraction; (3) fast mobility to facilitate an effective charge transfer; and (4) excellent chemical, optical, and ambient stability for a stable long-term perovskite solar cell.

From this perspective,  $\text{SnO}_2$  manifests as an excellent candidate for  $\text{MO}_x$ -based ETL in regular planar PSC due to its distinctive benefits. Additionally, general methodologies for  $\text{MO}_x$  optimization such as surface modification, bilayer engineering, composite development, and elemental doping are beneficial in adjusting the  $\text{MO}_x$  optoelectronic, chemical, and physical characteristics. As a result, it is possible to significantly advance perovskite solar cell technology toward better efficiency and high stability.

Notably, it is an inexpensive and low-energy technique to directly deposit ETLs using nanocrystal solution among the many methods used in producing  $\text{MO}_x$  ETLs. From this perspective, preparing and using colloidal nanocrystals can be useful in developing the lower-temperature ETLs of a planar perovskite solar cell. These colloidal nanocrystals have outstanding crystallinity and tunable optoelectronic properties. As a result, the ongoing research into innovative  $\text{MO}_x$  and process optimization for nanocrystal ETL will be a key component in the development of high-efficiency, inexpensive, large-area, and flexible devices. To boost the flexibility of ETL, mixing  $\text{MO}_x$  with functional organic compounds may therefore be a successful strategy. The organic molecules can strongly adsorb on the  $\text{MO}_x$  surface, particularly when certain functional groups are interacting with  $\text{MO}_x$ . Furthermore, by producing an organic network structure, the addition of chemically cross-linked organic elements can change a brittle  $\text{MO}_x$  ETL using a mechanically robust semiconducting structure. The surface modification of  $\text{MO}_x$  ETLs may help to increase flexibility by employing this method.

## Future Perspectives

Despite significant advancements over the past ten years, further research is still required for low-temperature solution-processed  $\text{MO}_x$  ETLs. The quest to explore novel electron transport materials with fast mobility, fewer surface imperfections, and suitable alignment of energy bands is an endless process. One of the main constraints impacting the  $\text{MO}_x$  ETL in flexible PSC is the intrinsically brittle nature of  $\text{MO}_x$ . Organics; in contrast, exhibit the virtue of good adaptability. The organic modification can also be used to adjust the  $\text{MO}_x$  surface energy, work function, and wetting characteristics. Owing to the several surface and interfacial cracks in perovskite, various functional groups, including Lewis acid and base, can be added to organic molecules to produce unique linkages with the perovskite layer. These connections can promote interfacial charge transfer, passivate trap states, and reduce the interfacial trap densities, thus improving the performance of the device and lowering hysteresis. All things considered, we believe that through the development and optimization of the  $\text{MO}_x$  in conjunction

with a thorough understanding of material science, the planar PSC utilizing low-temperature MO<sub>x</sub> ETLs has the potential to be a viable commercialization contender in the future due to its significant operational improvements.

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